

New transformation of α -acetylenic ketones under the action of 1,2-diaminoethane

M. P. Davydova,^a S. F. Vasilevskii,^{a*} and G. A. Tolstikov^b

^aInstitute of Chemical Kinetics and Combustion, Siberian Branch of the Russian Academy of Sciences, 3 ul. Institutskaya, 630090 Novosibirsk, Russian Federation.

Fax: +7 (383) 330 7350. E-mail: vasilev@kinetics.nsc.ru

^bN. N. Vorozhtsov Institute of Organic Chemistry, Siberian Branch of the Russian Academy of Sciences, 9 prosp. Akad. Lavrent'eva, 630090 Novosibirsk, Russian Federation

A reaction of 3-aryl-1-(3,4,5-trimethoxyphenyl)prop-2-yn-1-ones with 1,2-diaminoethane in boiling dioxane leads to 2-aryl-4,5-dihydro-1*H*-imidazoles and 1-(3,4,5-trimethoxyphenyl)ethanone.

Key words: α -acetylenic ketones, heterocyclization, 1,2-diaminoethane, imidazolidines.

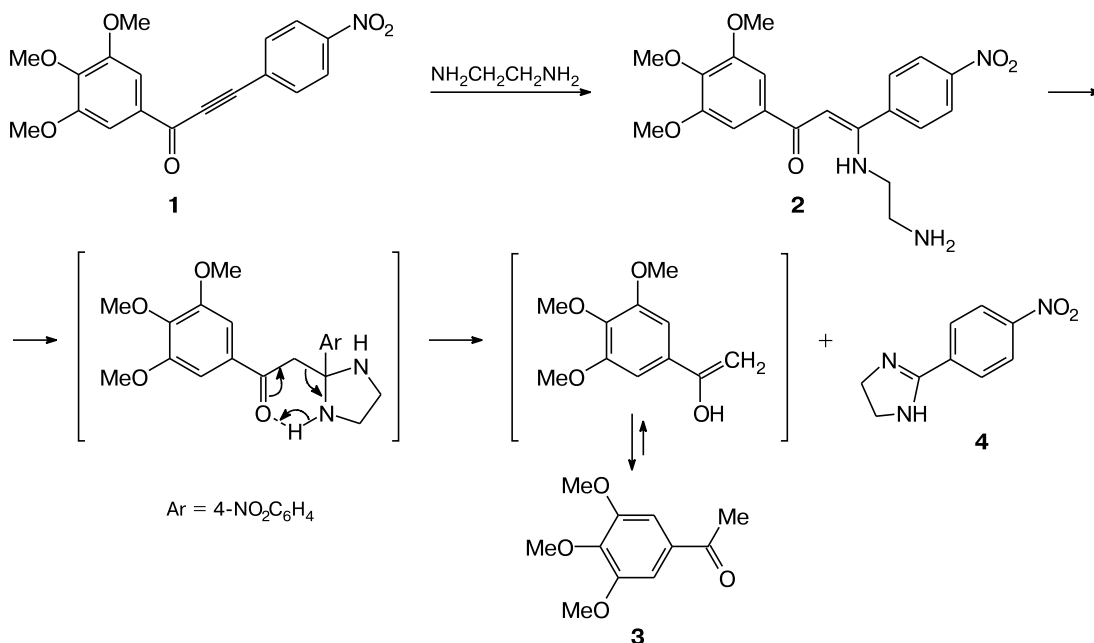
Earlier,¹ we have reported the data on the reaction of α -acetylenic ketone **1** with amino compounds (hydrazine hydrate, guanidine, hydroxylamine), which leads to heterocyclic derivatives, analogs of combretastatin A-4 (see Ref. 2). In continuation of our study of nucleophilic addition at the triple bond having in mind new analogs with a heterocyclic bridge, we studied a reaction of α -alkynylketones with 1,2-diaminoethane.

Herein we have found that reflux of 3-(4-nitrophenyl)-1-(3,4,5-trimethoxyphenyl)prop-2-yn-1-one (**1**)

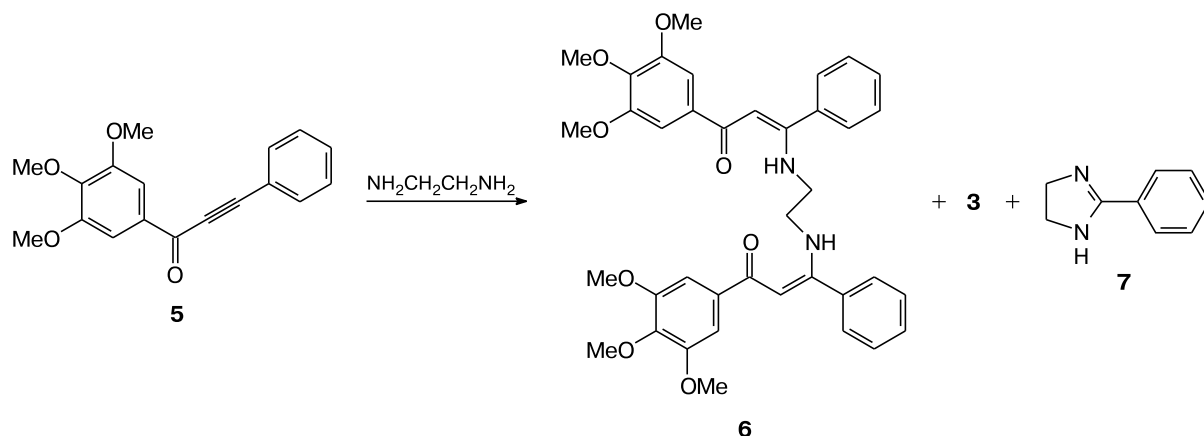
and 1,2-diaminoethane in dioxane (0.5 h) leads not only to the expected 3-(2-aminoethylamino)-3-(4-nitrophenyl)-1-(3,4,5-trimethoxyphenyl)prop-2-en-1-one (**2**) (13%), but also to unexpected products of fragmentation at the triple bond, *i.e.*, 1-(3,4,5-trimethoxyphenyl)ethanone (**3**) (35%) and 2-(4-nitrophenyl)-4,5-dihydro-1*H*-imidazole (**4**) (34%) (Scheme 1).

To find whether this reaction takes place in the case of substrates having no additional electron-withdrawing substituent (NO₂), we carried out the reaction of 1,2-diami-

Scheme 1



Scheme 2



noethane with phenylethynyl ketone **5** (Scheme 2). However, in this case formation of fragmentation products, ketone **3** and imidazole **7**, also took place, whose yields were 45 and 42%, respectively. In addition, instead of monosubstituted 1,2-diaminoethane, similar to compound **2**, a product of the double addition, symmetric diamine **6**, was formed (6%).

In conclusion, an unusually ready cleavage at the activated triple bond under the action of 1,2-diaminoethane was found, which leads to the corresponding aryl methyl ketones and 2-aryl-4,5-dihydro-1H-imidazoles.

Experimental

IR spectra were recorded on a Bruker Vector 22 spectrometer in KBr pellets. NMR spectra were recorded on a Bruker AV 400 spectrometer (400.13 MHz) in CDCl_3 or $\text{DMSO}-d_6$. Mass spectra were obtained on a Thermo Scientific DFS instrument (Double Focusing Sector Mass Spectrometer, Thermo Electron Corporation) with the direct inlet of the sample (the temperature of the ionization chamber was 220–270 °C, the ionization potential was 70 eV). Merck 60 (0.063–0.2 mm) silica gel was used for chromatography, TLC monitoring was performed on Silufol 60 F254 plates.

Reaction of 3-(4-nitrophenyl)-1-(3,4,5-trimethoxyphenyl)prop-2-yn-1-one (1) with 1,2-diaminoethane. A mixture of ketone **1** (see Ref. 1) (0.682 g, 2 mmol) and 1,2-diaminoethane (0.12 g, 2 mmol) in dioxane (10 mL) was refluxed for 0.5 h. The reaction progress was monitored by TLC (Silufol, AcOEt). The reaction mixture was concentrated *in vacuo*, the residue was subjected to chromatography on silica gel (eluent: hexane, hexane–toluene (1 : 1), toluene, toluene–AcOEt (1 : 1), AcOEt).

3-(2-Aminoethylamino)-3-(4-nitrophenyl)-1-(3,4,5-trimethoxyphenyl)prop-2-en-1-one (2). The yield was 0.11 g (13%), m.p. 258–260 °C (dioxane). ^1H NMR (CDCl_3), δ : 1.60 (br.s, 2 H, NH_2); 3.34 (m, 2 H, $\text{C}(1)\text{H}_2$); 3.76 (m, 2 H, $\text{C}(2)\text{H}_2$); 3.93 (s, 9 H, OMe); 5.73 (s, 1 H, =CH); 7.15 (s, 2 H, $\text{H}(2')$, $\text{H}(6')$); 7.53 (d, 2 H, $\text{H}(2)_{\text{Ph}}$, $\text{H}(6)_{\text{Ph}}$, $J = 8.6$ Hz); 8.29 (d, 2 H, $\text{H}(3)_{\text{Ph}}$,

$\text{H}(5)_{\text{Ph}}$, $J = 8.6$ Hz); 11.13 (s, 1 H, NH). ^{13}C NMR, δ : 45.25 (CH_2NH_2); 56.12 (2 OCH₃); 60.83 (OCH₃); 67.84 (CH_2NH); 94.03 (CH); 104.34 ($\text{C}(2')$, $\text{C}(6')$); 124.02 ($\text{C}(3)$, $\text{C}(5)$); 128.87 ($\text{C}(2)$, $\text{C}(6)$); 134.40 ($\text{C}(1')$); 140.85 ($\text{C}(1)$); 141.26 ($\text{C}(4')$); 148.33 ($\text{C}(4)$); 153.03 ($\text{C}(3')$, $\text{C}(5')$); 163.67 (CNH); 188.06 (CO). IR (KBr), ν/cm^{-1} : 3429 (NH). HRMS, found: m/z 384.1313 [$\text{M} - \text{NH}_3$]⁺ ($\text{C}_{20}\text{H}_{20}\text{O}_6\text{N}_2$), 357.1084 [$\text{M} - \text{CH}_3\text{CH}=\text{NH}_2$]⁺ ($\text{C}_{18}\text{H}_{17}\text{O}_6\text{N}_2$). Calculated: $\text{M} = 401.1581$ ($\text{C}_{20}\text{H}_{23}\text{N}_3\text{O}_6$).

The reaction also yielded **1-(3,4,5-trimethoxyphenyl)ethanone (3)** (0.15 g, 35%), m.p. 76–78 °C (benzene) (cf. Ref. 3: m.p. 77–79 °C) and **2-(4-nitrophenyl)-4,5-dihydro-1H-imidazole (4)** (0.13 g, 34%), m.p. 242–244 °C (dioxane) (cf. Ref. 4: m.p. 242–243 °C).

The reaction of 1-(3,4,5-trimethoxyphenyl)-3-phenylprop-2-yn-1-one (**5**)⁵ with 1,2-diaminoethane was carried out similarly, the reaction time was 9 h. The reaction mixture was concentrated *in vacuo*, the residue was subjected to chromatography on silica gel (eluent hexane, hexane–toluene (1 : 1), toluene, toluene–AcOEt (1 : 1), AcOEt).

N,N'-Bis[3-oxo-1-phenyl-3-(3,4,5-trimethoxyphenyl)prop-1-en-1-yl]ethane-1,2-diamine (6). The yield was 0.07 g (6%), m.p. 202–204 °C (benzene). ^1H NMR (CDCl_3), δ : 3.33 (m, 2 H, CH_2); 3.88 (s, 9 H, OMe); 5.68 (s, 1 H, =CH); 7.12 (s, 2 H, $\text{H}(2')$, $\text{H}(6')$); 7.26 (m, 2 H, $\text{H}(2)_{\text{Ph}}$, $\text{H}(6)_{\text{Ph}}$); 7.44 (m, 3 H, $\text{H}(3)_{\text{Ph}}$ – $\text{H}(5)_{\text{Ph}}$); 11.26 (s, 1 H, NH). ^{13}C NMR, δ : 44.97 (CH_2); 56.10 (2 OCH₃); 60.78 (OCH₃); 93.82 (CH); 104.22 ($\text{C}(2')$, $\text{C}(6')$); 127.51 ($\text{C}(2)$, $\text{C}(6)$); 128.63 ($\text{C}(3')$, $\text{C}(5')$); 129.41 ($\text{C}(4)$); 134.99 ($\text{C}(1')$); 135.35 ($\text{C}(1)$); 140.54 ($\text{C}(4')$); 152.81 ($\text{C}(3')$, $\text{C}(5')$); 166.48 (CNH); 187.67 (CO). IR (KBr), ν/cm^{-1} : 3437 (NH). Found (%): C, 69.94; H, 6.17; N, 4.35. $\text{C}_{38}\text{H}_{40}\text{N}_2\text{O}_8$. Calculated (%): C, 69.96; H, 6.18; N, 4.29.

The reaction also yielded **1-(3,4,5-trimethoxyphenyl)ethanone (3)** (0.35 g, 45%), m.p. 76–78 °C (benzene) (cf. Ref. 3: m.p. 77–79 °C) and **2-phenyl-4,5-dihydro-1H-imidazole (7)** (0.21 g, 42%), m.p. 93–95 °C (benzene) (cf. Ref. 6: m.p. 92–94 °C).

This work was financially supported by the Siberian Branch of the Russian Academy of Sciences (Interdisciplinary Grant 93), the Russian Foundation for Basic Research (Project No. 10-03-00257-a), the Russian Academy of Sciences (Grant 5.9.3), and the Chemistry Service

Center of the Siberian Branch of the Russian Academy of Sciences.

References

1. S. F. Vasilevskii, M. P. Davydova, G. A. Tolstikov, *Khim. Geterotsikl. Soedin.*, 2008, 1545 [*Chem. Heterocycl. Compd. (Engl. Transl.)*, 2008, **44**, 1257].
2. G. C. Tron, T. Pirali, G. Sorba, F. Pagliali, S. Busacca, A. Genazzani, *J. Med. Chem.*, 2006, **49**, 3033.
3. V. J. Harding, *J. Chem. Soc. London*, 1914, **105**, 2790.
4. V. B. Piskov, V. P. Kasperovich, L. M. Yakovleva, *Khim. Geterotsikl. Soedin.*, 1976, 1112 [*Chem. Heterocycl. Compd. (Engl. Transl.)*, 1976, **12**, 917].
5. M. L. Edwards, D. M. Stemerick, P. S. Sunkara, *J. Med. Chem.*, 1990, **33**, 1948.
6. H. Prokopcov, C. O. Kappe, *J. Org. Chem.*, 2007, **72**, 4440.

Received June 30, 2010;
in revised form October 19, 2010